

## Lampiran 1. Pembuatan Reagen dan Desain Primer

### 1. Pembuatan Buffer Lisis

Larutan *stock* Tris-Cl 1 M pH 8, EDTA 2-Na 0.5 M pH 8 dan NaCl 5 M masing-masing sebanyak 30 mL serta SDS 10% sebanyak 20 mL.

Tris-Cl 1 M pH 8 30 ml (0.03 liter); Mr Tris-Cl (121.14) :

$$\begin{aligned}\text{Gram} &= M \times \text{Mr} \times V \\ &= 1 \times 121.14 \times 0.03 \\ &= 3.6342 \text{ gram Tris-Cl} + 30 \text{ ml dH}_2\text{O} \text{ \& diadjust pH} \\ &\text{hingga 8}\end{aligned}$$

EDTA 2-Na 0.5 M pH 8 (0.03 liter); Mr EDTA 2-Na (372.24)

$$\begin{aligned}\text{Gram} &= M \times \text{Mr} \times V \\ &= 0.5 \times 372.24 \times 0.03 \\ &= 5.5836 \text{ gram EDTA 2-Na} + 30 \text{ ml dH}_2\text{O} \text{ \&} \\ &\text{diadjust pH hingga 8}\end{aligned}$$

NaCl 5 M 30 mL (0.03 liter); Mr NaCl (58.44)

$$\begin{aligned}\text{Gram} &= M \times \text{Mr} \times V \\ &= 5 \times 58.44 \times 0.03 \\ &= 8.766 \text{ gram NaCl} + 30 \text{ ml dH}_2\text{O}\end{aligned}$$

SDS 10% 20 mL (0.02 liter); Mr SDS (288.38)

$$10\% \times 20 \text{ mL} = 2 \text{ ml (jika dalam bentuk liquid)} + 20 \text{ ml dH}_2\text{O}$$

$$10\% \times 20 \text{ mL} = 2 \text{ gram (jika dalam bentuk serbuk)} + 20 \text{ mL dH}_2\text{O}$$

Buffer lysis (Tris-Cl 10 mM, EDTA 2-Na 25 mM, NaCl 100 mM, SDS 1% & Pro-K) 12 mL (untuk 8 sampel sebanyak 3 ulangan)

Tris-Cl 10 mM

$$\begin{aligned}M_1 \times V_1 &= M_2 \times V_2 \\ 10 \text{ mM} \times 12 \text{ mL} &= 1 \text{ M} \times X \\ X &= 0,12 \text{ ml (120 } \mu\text{l) Tris-Cl 1 M pH 8} \\ &\text{dari stock}\end{aligned}$$

EDTA 2-Na 25 mM

$$\begin{aligned}
 M_1 \times V_1 &= M_2 \times V_2 \\
 25 \text{ mM} \times 12 \text{ ml} &= 0.5 \text{ M} \times X \\
 X &= 0,6 \text{ mL (600 } \mu\text{l) EDTA 2-Na 0.5} \\
 &\text{ M pH 8 dari } \textit{stock}
 \end{aligned}$$

$$\begin{aligned}
 \text{NaCl 100 mM} \\
 M_1 \times V_1 &= M_2 \times V_2 \\
 100 \text{ mM} \times 12 \text{ ml} &= 5 \text{ M} \times X \\
 X &= 0,24 \text{ ml (240 } \mu\text{l) NaCl 5 M dari} \\
 &\textit{stock}
 \end{aligned}$$

$$\begin{aligned}
 \text{SDS 1\%} \\
 M_1 \times V_1 &= M_2 \times V_2 \\
 1\% \times 12 \text{ ml} &= 10\% \times X \\
 X &= 1,2 \text{ ml (1200 } \mu\text{l) SDS 10\% dari} \\
 &\textit{stock}
 \end{aligned}$$

$$\begin{aligned}
 \text{Proteinase-K (Pro-K)} \\
 \text{Stock} &= 20 \text{ mg/ml} \\
 \text{Pro-K : buffer lysis} &= 1 : 40 \\
 &= 300 : 12000 \text{ (untuk 10 mg/ml)} \\
 &= 150 : 12000 \text{ (untuk 20 mg/ml)}
 \end{aligned}$$

## 2. Pembuatan PCI (*Phenol : Chloroform : Isoamyl Alcohol*) (25:24:1) untuk 1 ml (1000 $\mu$ l)

$$\begin{aligned}
 \text{Phenol} &= \frac{25}{50} \times 1000 \\
 &= 500 \mu\text{l}
 \end{aligned}$$

$$\begin{aligned}
 \text{Chloroform} &= \frac{24}{50} \times 1000 \\
 &= 480 \mu\text{l}
 \end{aligned}$$

$$\begin{aligned}
 \text{Isoamyl Alcohol} &= \frac{1}{50} \times 1000 \\
 &= 20 \mu\text{l}
 \end{aligned}$$

## 3. Pembuatan NaOAc 3 M pH 5.2 30 ml (0.03 liter); Mr NaOAc (82.03)

$$\text{Gram} = M \times \text{Mr} \times V$$

$$= 3 \times 82.03 \times 0.03$$

$$= 7.3827 \text{ gram NaOAc} + 30 \text{ mL dH}_2\text{O} \text{ \& diadjust}$$

pH hingga 5.2

#### 4. Pembuatan Buffer TE pH 7.6 20 mL (10 mM Tris-Cl & 1 mM EDTA 2-Na)

*Stock*

Buffer TE pH 7.6

Tris-Cl 1 M pH 8

EDTA 2-Na 0.5 M pH 8

dH<sub>2</sub>O

Tris-Cl 10 mM pH 8

EDTA 2-Na 1 mM pH 8

dH<sub>2</sub>O lalu diadjust pH

Tris-Cl 10 mM pH 8

$M_1 \times V_1$

10 mM x 20 ml

X

$= M_2 \times V_2$

$= 1 \text{ M} \times X$

$= 0.2 \text{ mL (200 } \mu\text{l) Tris-Cl 1 M pH 8}$   
dari *stock*

EDTA 2-Na 1 mM pH 8

$M_1 \times V_1$

1 mM x 20 ml

X

$= M_2 \times V_2$

$= 0.5 \text{ M} \times X$

$= 0.04 \text{ mL (40 } \mu\text{l) EDTA 2-Na 0.5}$   
 $\text{M pH 8 dari } \textit{stock}$

Ditambah dengan dH<sub>2</sub>O hingga volumenya menjadi 20 ml lalu diadjust pH hingga 7.6

#### 5. Pembuatan Buffer TBE 10x pH8 500 ml (0,05 M Tris-Cl; 0,05 M Asam Boric; 0,01M EDTA 2-Na)

Tris-Cl (Mr 121,14)

gr

gr

gr

Boric acid (Mr 61,83)

gr

gr

gr

$= M \times V \times Mr$

$= 0,05 \text{ M} \times 0,5 \text{ L} \times 121,14$

$= 3,035 \text{ gr (1x)} = 30,3 \text{ (10x)}$

$= M \times V \times Mr$

$= 0,05 \text{ M} \times 0,5 \text{ L} \times 61,83$

$= 1,54 \text{ gr (1x)} = 15,4 \text{ gr (10x)}$

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CPU: 100%  
MEM: 100%  
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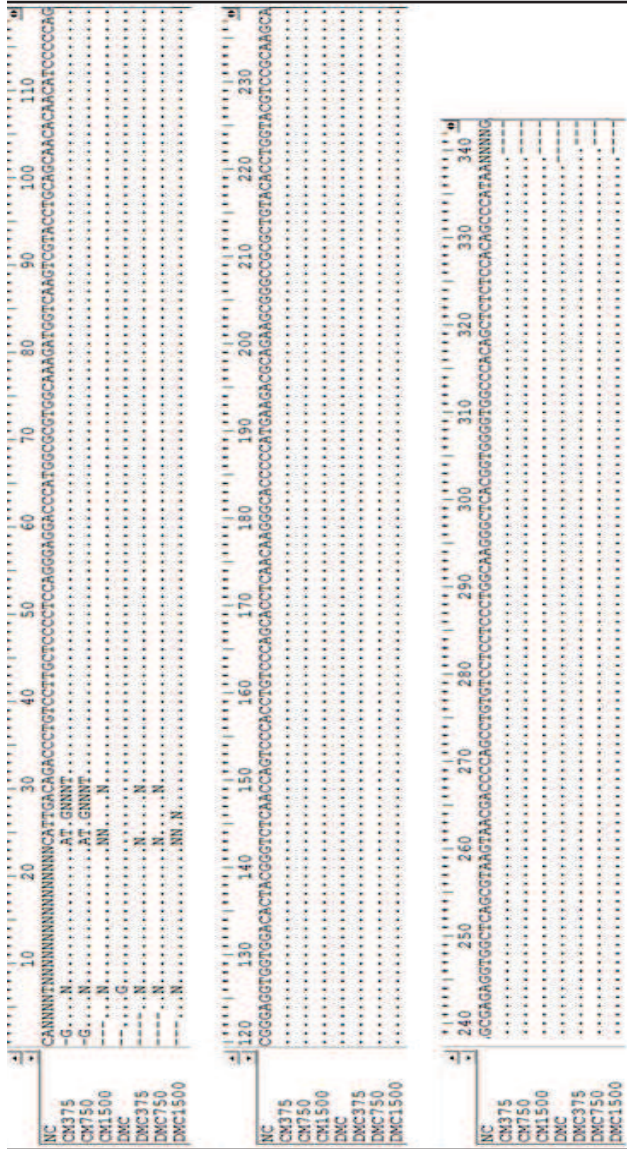
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## 9. Sekuen Exon 4 Gen *Hnfla*

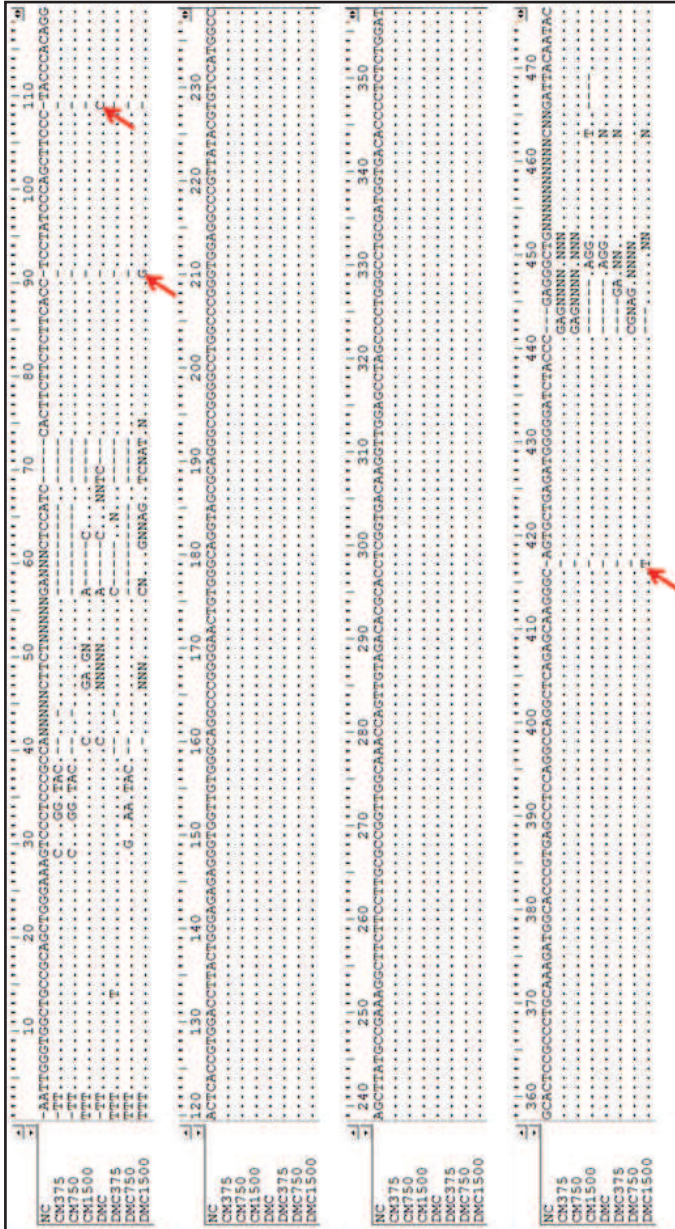
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tccagccctc  
16921 gggtagatcc cccatctcag cactgccctt gctctgagcc tggcctggag  
gctcacgggt  
16981 gccatctttg cagggeggag tgcatecaga gaggggtgtc accatcagcag  
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tgaagagaag  
17281 aagtggatgg gtagaaggc gggagggact ttcccagctg cggcagccac  
ccaaaaatg

Lampiran 2. Hasil Penyejajaran Sekuen gen *Hnf1a*



Gambar 10. Gen *Hnf1a* exon 2



Gambar 11. Gen *Hmfl*  $\alpha$  exon 4